

# **Carbon Dioxide**

# **Safety Data Sheet**

# SECTION 1: Identification of the substance/mixture and of the company/undertaking

**Product identifier** 

Product Name Carbon Dioxide

Other means of identification

Safety data sheet number SDS-YO-007 UN/ID no. UN1013

Trade name Carbon Dioxide, CO2 Gas, CO2 N2.7, CO2 HP/Medical N3.0

Recommended use of the chemical and restrictions on use

Recommended Use Industrial and professional use. Food and Beverage

Uses advised against Consumer use

#### Details of the supplier of the safety data sheet

Yateem Oxygen W.L.L

P.O. Box 60, Manama, Bahrain

Email: wecare@yateemoxygen.com Website: www.yateemoxygen.com Customer Service: +973 17400677

#### **Emergency telephone number**

Company Phone Number +973 17400456

Emergency Contact Number +973 17456248; +973 17400675

# **SECTION 2: Hazards identification**

# Classification

**OSHA Regulatory Status** 

This chemical is considered hazardous by the 2012 OSHA Hazard Communication Standard (29 CFR 1910.1200).

This enemical is considered nazaraodo by the 2012 Contri trazara Contint	diffication Standard (20 Of It 1010:1200)
Gasses under pressure	Liquified Gas
Simple asphyxiants	Yes

# Label elements

Signal word Warning

#### **Hazard Statements**

Contains gas under pressure; may explode if heated May displace oxygen and cause rapid suffocation

May cause frostbite

May increase respiration and heart rate

#### **Precautionary Statements - Prevention**

Do not handle until all safety precautions have been read and understood

Avoid breathing gas

Do not get in eyes, on skin, or on clothing

Use and store only outdoors or in a well-ventilated place

Use a backflow preventive device in piping

Use only with equipment rated for cylinder pressure

Close valve after each use and when empty

# **Precautionary Statements - Response:**

IF INHALED: Remove person to fresh air and keep comfortable for breathing. Get medical attention/advice.

IF ON SKIN: Get immediate medical advice/attention. Thaw frosted parts with lukewarm water. Do not rub affected area.

Precautionary Statements – Storage: Protect from sunlight when ambient temperature exceeds 52°C/125°F

Hazards not otherwise classified (HNOC): Not applicable



# SECTION 3: Composition/information on ingredients

Chemical Name	CAS No.	Volume %	Chemical Formula	
CARBON DIOXIDE	124-38-9	>99	CO <sub>2</sub>	

## **SECTION 4: First aid measures**

#### Description of first aid measures

**General advice** Show this safety data sheet to the doctor in attendance.

Inhalation Remove to fresh air and keep comfortable for breathing. If breathing is difficult, give oxygen. If breathing

has stopped, give artificial respiration. Get medical attention immediately

**Skin contact** For dermal contact or suspected frostbite, remove contaminated clothing and flush affected areas with

lukewarm water. DO NOT USE HOT WATER. A physician should see the patient promptly if contact

with the product has resulted in blistering of the dermal surface or in deep tissue freezing.

Eye contact If frostbite is suspected, flush eyes with cool water for 15 minutes and obtain immediate medical

attention.

**Ingestion** Not an expected route of exposure

Self-protection of the first aider Rescue personnel should be equipped WITH SELF-CONTAINED BREATHING APPARATUS

#### Most important symptoms and effects, both acute and delayed

Simple asphyxiant. May cause suffocation by displacing the oxygen in the air. Exposure to oxygen-deficient atmosphere (<19.5%) may cause dizziness, drowsiness, nausea, vomiting, excess salivation, diminished mental alertness, loss of consciousness and death. Exposure to atmospheres containing 8-10% or less oxygen will bring about unconsciousness without warning and so quickly that the individuals cannot help or protect themselves. Lack of sufficient oxygen may cause serious injury or death. Depending on concentration and duration of exposure to carbon dioxide may cause increased respirations, headache, mild narcotic effects, increased blood pressure and pulse, and asphyxiation. Symptoms of overexposure become more apparent when atmospheric oxygen is decreased to 15-17%. Contact with evaporating liquid may cause cold burns/ frostbite.

# In Indication of any immediate medical attention and special treatment needed

Note to physicians Treat symptomatically.

#### SECTION 5: Firefighting measures

#### Suitable extinguishing media

Use extinguishing measures that are appropriate to local circumstances and the surrounding environment.

Unsuitable extinguishing media None

# Specific extinguishing methods

Continue to cool fire exposed cylinders until flames are extinguished. Damaged cylinders should be handled only by specialists.

#### Specific hazards arising from the chemical

Non-flammable gas. Cylinders may rupture under extreme heat.

#### Protective equipment and precautions for firefighters

As in any fire, wear self-contained breathing apparatus pressure-demand, NIOSH (approved or equivalent) and full protective gear.

#### **SECTION 6: Accidental release measures**

# Personal precautions, protective equipment and emergency procedures

**Personal precautions** Evacuate personnel to safe areas. Ensure adequate ventilation, especially in confined areas. Monitor

oxygen level. Wear self-contained breathing apparatus when entering area unless atmosphere is proved to

be safe

Other information Gas/vapor is heavier than air. Prevent from entering sewers, basements and work pits, or any

place where accumulation may be dangerous

**Environmental precautions** 

**Environmental precautions** Prevent spreading of vapors through sewers, ventilation systems and confined areas.

#### Methods and material for containment and cleaning up

Methods for containment Stop the flow of gas or remove cylinder to outdoor location if this can be done without risk. If leak is in

container or container valve, contact the appropriate emergency telephone number in Section 1 or call your closest Yateem Oxygen location. DO NOT ATTEMPT TO REMOVE CYLINDERS THAT HAVE BEEN

EXPOSED TO HEAT.

Return cylinder to Yateem Oxygen Methods for cleaning up

# SECTION 7. Handling and Storage

# Precautions for safe handling Advice on safe handling

For applications with moist Carbon Dioxide, 316, 309 and 310 stainless steels may be used as well as Hastelloy® A, B, & C and Monel®. Ferrous nickel alloys are slightly suspectable to corrosion. At normal temperatures carbon dioxide is compatible with most plastics and elastomers

Protect cylinders from physical damage; do not drag, roll, slide or drop. When moving cylinders, even for short distance, use a cart designed to transport cylinders. Never attempt to lift a cylinder by its valve protection cap. Never insert an object (e.g. wrench, screwdriver, pry bar,etc.) into valve cap openings. Doing so may damage valve, causing leak to occur. Use an adjustable strap wrench to remove over-tight or rusted caps. Use only with adequate ventilation. Use a backflow preventive device in piping. Use only with equipment rated for cylinder pressure. Close valve after each use and when empty. If user experiences any difficulty operating cylinder valve discontinue use and contact supplier. Ensure the complete gas system has been checked for leaks before use.

Never put cylinders into trunks of cars or unventilated areas of passenger vehicles. Never attempt to refill a compressed gas cylinder without the owner's written consent. Never strike an arc on a compressed gas cylinder or make a cylinder a part of an electrical circuit.

Only experienced and properly instructed persons should handle gases under pressure. Always store and handle compressed gas cylinders in accordance with Compressed Gas Association, pamphlet CGA-P1, Safe Handling of Compressed Gases in Containers.

For additional recommendations, consult Compressed Gas Association's Pamphlets SB-7, G-4.3, G-4.1, G-4.4, P-2.5, G-4.9, P-14, and SB-2

# Conditions for safe storage, including any incompatibilities

**Storage Conditions** Store in cool, dry, well-ventilated area of non-combustible construction away from heavily

trafficked areas and emergency exits. Keep at temperatures below 52°C / 125°F. Cylinders should be stored upright with valve protection cap in place and firmly secured to prevent falling. Full and empty cylinders should be segregated. Use a "first in-first out" inventory system to prevent full cylinders from being stored for excessive periods of time. Stored containers should be periodically checked for general condition and leakage

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Certain reactive metals, hydrides, moist cesium monoxide, or lithium acetylene carbide diammino Incompatible materials may ignite. Passing carbon dioxide over a mixture of sodium peroxide and aluminum or

magnesium may explode

# SECTION 8: Exposure controls/personal protection

#### **Control parameters**

Chemical Name	ACGIH TLV	OSHA PEL	NIOSH IDLH
CARBPM DIOXIDE	STEL: 30000 ppm	TWA: 5000 ppm	IDLH: 40000 ppm
124-38-9	TWA: 5000 ppm	TWA: 9000 mg/m3 (vacated) TWA: 10000 ppm (vacated) TWA: 18000 mg/m3 (vacated) STEL: 30000 ppm (vacated) STEL: 54000 mg/m3	TWA: 5000 ppm TWA: 9000 mg/m3 STEL: 30000 ppm STEL: 54000 mg/m3

ACGIH TLV: American Conference of Governmental Industrial Hygienists - Threshold Limit Value. OSHA PEL: Occupational Safety and Health Administration - Permissible Exposure Limits. NIOSH IDLH: Immediately Dangerous to Life or Health.

Other Information Vacated limits revoked by the Court of Appeals decision in AFL-CIO v. OSHA, 965 F.2d 962

(11th Cir., 1992).

**Appropriate Engineering Controls** 

Engineering Controls Local exhaust ventilation to prevent accumulation of high concentrations and maintain air-

oxygen levels at or above 19.5%. Oxygen detectors should be used when asphyxiating gases may be released. Systems under pressure should be regularly checked for

leakages.

Individual protection measures, such as personal protective equipment

**Eye/face protection** Wear safety glasses with side shields (or goggles).

If splashes are likely to occur, wear: Goggles. Face-shield.

**Skin and body protection**Work gloves and safety shoes are recommended when handling cylinders.

Wear cold insulating gloves when handling liquid

**Respiratory protection**None under normal use. Use positive pressure airline respirator with escape cylinder or

self-contained breathing apparatus for oxygen-deficient atmospheres (<19.5%). If exposure limits are exceeded or irritation is experienced, NIOSH approved respiratory protection should be worn. Positive pressure supplied air respirators may be required for high airborne contaminant concentrations. Respiratory protection must be provided in

accordance with current local regulations

General Hygiene Considerations Handle in accordance with good industrial hygiene and safety practice.

Do not get in eyes, on skin, or on clothing

#### **SECTION 9: Physical and chemical properties**

# Information on basic physical and chemical properties

Physical stateGasAppearanceColorless.OdorOdorless

Odor threshold No information available

pH Not applicable

Melting/freezing point -56.6 °C / -69.8 °F

Evaporation rate Not applicable

Flammability (solid, gas) Non-Flammable gas

Lower flammability limit: Not applicable

Upper flammability limit: Not applicable

Flash point Not applicable

Autoignition temperatureNo data availableDecomposition temperatureNo data availableWater solubility0.145 g/ml @25 °CPartition coefficientNo data availableKinematic viscosityNot applicable

Chemical Name	Molecular weight	Boiling point / range	Vapor Pressure	Vapor density (air =1)	Gas Density kg/m³@20°C	Critical Temperature
CARBON DIOXIE	44.01	-78.5 °C (sublimes)	57780hPa @21.1 °C	1.522	1.839	31.1°C

# **SECTION 9: Stability and Reactivity**

#### Reactivity

Not reactive under normal conditions

#### **Chemical stability**

Stable under normal conditions.

#### **Explosion data**

Sensitivity to Mechanical Impact None Sensitivity to Static Discharge None

#### Possibility of Hazardous Reactions

None under normal processing.

#### **Conditions to avoid**

Due to the presence of Carbon dioxide, Carbonic acid is formed in the presence of moisture

#### **Incompatible materials**

Certain reactive metals, hydrides, moist cesium monoxide, or lithium acetylene carbide diamino may ignite. Passing carbon dioxide over a mixture of sodium peroxide and aluminum or magnesium may explode.

#### **Hazardous Decomposition Products**

Oxygen. Carbon monoxide.

# **SECTION 11: Toxicological Information**

#### Information on likely routes of exposure

**Inhalation** Acidosis, adrenal cortical exhaustion, and other metabolic stresses have resulted from prolonged

continuous exposure to 1-2% carbon dioxide (10,000 ppm-20,000 ppm). The ACGIH TLV of 5,000 ppm is expected to provide a good margin of safety from asphyxiation and undue metabolic stress provided sufficient oxygen levels are maintained in the air. Increased physical activity, duration of exposure, and decreased oxygen content can affect systemic and respiratory effects resulting from

exposure to carbon dioxide

Skin contact

Contact with evaporating liquid may cause cold burns/frostbite.

Eye contact

Contact with evaporating liquid may cause cold burns/frostbite.

**Ingestion** Not an expected route of exposure.

#### Information on toxicological effects

**Symptoms** Depending on concentration and duration of exposure to carbon dioxide may cause

increased respirations, headache, mild narcotic effects, increased blood pressure and pulse, and asphyxiation. Symptoms of overexposure become more apparent when atmospheric

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oxygen is decreased to 15-17%.

#### Delayed and immediate effects as well as chronic effects from short and long-term exposure

IrritationNot classified.SensitizationNot classified.Germ cell mutagenicityNot classified.

**Carcinogenicity** It does not contain any carcinogens or potential carcinogens listed by OSHA, IARC or NTP.

Reproductive toxicity

Developmental Toxicity

STOT - single exposure

STOT - repeated exposure

Not classified.

Not classified.

Chronic toxicity Chronic harmful effects are not known from repeated inhalation of concentrations below

PEL/TLV.

Target Organ Effects Central Vascular System (CVS), Respiratory system.

Aspiration hazard Not applicable.

**Numerical measures of toxicity** 

Chemical Name	Oral LD50	Dermal LD50	Inhalation LC50	Inhalation LC50 (CGA P20)
CARBON DIOXIDE 124-38-9	-	-	47,000 ppm (Rat)	-

Inhalation LC50 TCLo - 10,000 ppm (Rat) 24 hours/30 days-continuous

# **SECTION 12: Ecological Information**

#### **Ecotoxicity**

No known acute aquatic toxicity.

## Persistence and degradability

No information available

# **Bioaccumulation**

No information available

Global warming potential (GWP)

# **SECTION 13. Disposal Considerations**

#### Waste treatment methods

**Disposal of wastes**Do not attempt to dispose of residual waste or unused quantities. Return in the shipping container

PROPERLY LABELED WITH ANY VALVE OUTLET PLUGS OR CAPS SECURED AND VALVE

PROTECTION CAP IN PLACE to YATEEM OXYGEN for proper disposal.

# SECTION 14. Transportation Information

DOT

**UN/ID no.** UN1013

Proper shipping name Carbon Dioxide

Hazard Class 2.2 Subsidiary class 5.1

**Description** UN1013, Carbon Dioxide, 2.2

Emergency Response Guide Number 120

TDG

**UN/ID no.** UN1013

Proper shipping name Carbon Dioxide

Hazard Class 2.2

**Description** UN1013, Carbon Dioxide, 2.2

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IATA

UN/ID no. UN1013
Proper shipping name Carbon Dioxide

Hazard Class 2.2 ERG Code 2L

**Description** UN1013, Carbon Dioxide, 2.2

**IMDG** 

**UN/ID no.** UN1013

Proper shipping name Carbon Dioxide

Hazard Class 2.2 EmS-No. F-C, S-V

**Description** UN1013, Carbon Dioxide, 2.2

# **SECTION 15. Regulatory Information**

National Legislation Complies SEC <a href="https://www.sce.gov.bh/en/index">https://www.sce.gov.bh/en/index</a>

MTT <a href="http://www.transportation.gov.bh/content/caa-laws-and-regulations">http://www.transportation.gov.bh/content/caa-laws-and-regulations</a>

OHSC <a href="http://www.scosh.org/en/legislation/legislations#legislationContainer">http://www.scosh.org/en/legislation/legislationSeptimental http://www.scosh.org/en/legislation/legislationSeptimental http://www.scosh.org/en/legislation/legislationSeptimental http://www.scosh.org/en/legislation/legislationSeptimental http://www.scosh.org/en/legislation/legislationSeptimental http://www.scosh.org/en/legislationSeptimental http://www.scosh.org/

International Inventories

TSC :Complies DSL/NDSL :Complies EINECS/ELINCS :Complies

Legend

TSCA - United States Toxic Substances Control Act Section 8(b) Inventory DSL/NDSL - Canadian Domestic Substances List/Non-Domestic Substances List

EINECS/ELINCS - European Inventory of Existing Chemical Substances/European List of Notified Chemical Substances

SEC - Specific Council of Environment

MTT - Ministry of Transport and Telecommunications;

**OHSC** - Occupational Health and Safety Council

# **SECTION 16: Other Information**

NFPA Health hazards 2 Flammability 0 Instability 0 Physical and Chemical Properties

Simple asphyxiant

Note: Ratings were assigned in accordance with Compressed Gas Association (CGA) guidelines as published in CGA Pamphlet P-19-2009, CGA Recommended Hazard Ratings for Compressed Gases, 3rd Edition.

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Revision Note: SDS sections updated; 1, 6, 13 and 15

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**End of Safety Data Sheet**